0 1	Sodium hydroxide solution can be added to magnesium chloride solution. The formula equation for the reaction is shown below:
	$MgCl_{2(aq)} + 2NaOH_{(aq)} \to Mg(OH)_{2(s)} + 2NaCl_{(aq)}$
	This is an example of a precipitation reaction.
0 1 . 1	What is the meaning of the term precipitation reaction?  [2 marks]
0 1 . 2	What is the colour of this precipitate?  [1 mark]
0 1 . 3	How could the magnesium hydroxide, Mg(OH) <sub>2</sub> , be separated from the products of the reaction?  [1 mark]
0 2	Two students tested a green compound <b>X</b> .  The students added water to compound <b>X</b> . Compound <b>X</b> did not dissolve.  Student <b>A</b> carried out a flame test and observed an intense yellow flame.
	Student <b>B</b> added sodium hydroxide solution and a blue precipitate formed.  Student <b>A</b> concluded that compound <b>X</b> was a sodium compound. Student <b>B</b> concluded that compound <b>X</b> contained copper.
0 2 . 1	Which student, if any, was correct? Explain your reasoning.  [4 marks]
	(Total 8 marks)

End