Clean drinking water is important for health. In the UK water from reservoirs goes through a series of steps before it is safe to drink.
[2 marks]

What are the two mains steps used to treat water from reservoirs?
Give a reason for each step.

Step 1 $\qquad$

Reason $\qquad$

Step 2 $\qquad$

Reason $\qquad$

Explain why it is more difficult to produce drinking water from waste water than from water in lakes.
[3 marks]
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$\qquad$
a) Pure water can be produced by distillation.

Why is distillation not usually an economic method of treating water for drinking?
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$\qquad$
b) How could the water be tested to show it is pure? Give the expected result of the test for pure water.
$\qquad$
$\qquad$


A student wanted to find out how much solid was dissolved in sea water. This is the method the student used:

- measure the mass of an empty evaporating basin
- measure $25 \mathrm{~cm}^{3}$ of sea water and pour it into the evaporating basin
- heat the evaporating basin gently until all of the water has evaporated
- measure the mass of the evaporating basin containing the solid residue.
(a) What piece of apparatus would be suitable for measuring $25 \mathrm{~cm}^{3}$ of sea water?
(b) How could the student check that all of the water had evaporated?
[2 marks]
$\qquad$
$\qquad$

The results the student obtained using $25 \mathrm{~cm}^{3}$ of sea water are:
mass of empty evaporating basin $=23.21 \mathrm{~g}$ mass of evaporating basin and dry solid residue $=24.04 \mathrm{~g}$

Calculate the mass of solid dissolved in $1000 \mathrm{~cm}^{3}$ of the sea water.
$\qquad$
$\qquad$
Mass dissolved in $1000 \mathrm{~cm}^{3}=$ $\qquad$ g
(Total 13 marks)

## End

