0 1	Crude oil is	a mixture of hyd	rocarbons which can b	pe separated into fi	ractions.
0 1 . 1	What is a m	ixture?			
	Two or more	e elements or co	mpounds [1]		[2 marks]
	Not chemic	ally combined or	r joined together [1]		
0 1 . 2	What is a hy				[2 marks]
	A compound	d or molecule m	ade up of hydrogen [1]		
	and carbon	only. [1]			
0 2	The table be crude oil.	Boiling point	information on the hy  Relative % in crude	Carbon chain	s found in  Demand
	Namada	°C	oil	length	la i a la
	Naptha	125	10	5 to 10	high
	Kerosene	160	15	10 to 15	high
	Diesel	250	20	15 to 20	medium
	Fuel oil	300	45	30 to 70	low
0 2 . 1	Naphtha ha	s a lower boiling	point than diesel.		
	Explain why				fo
	Naptha (or i molecules o	t) contains shor or contains fewe	t or shorter chains/hydr r carbons.	drocarbons or sma [1]	[2 marks] ller
	Naptha (or i between the		smaller intermolecular	forces or smaller (	orces
0 2 . 2	Kerosene is	more expensive	than fuel oil.		
	Suggest rea	sons why.			[0
	Kerosene (c	or it) is in higher	or high demand.	[1]	[2 marks]
	It is in short	t supply.		[1]	

## WARNING:

Answers using the word 'it' will be given credit only if it is clear that the 'it' refers to the correct subject. So be careful with the use of the word 'it'!

0 3
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0 3 . 1

Crude oil is a mixture of mainly alkanes. Alkanes are a group of chemicals which contain carbon and hydrogen atoms, the smallest of which is methane.

Describe how a mixture of alkanes can be separated using fractional distillation.

[3 marks]

## NOTE:

Examiners report how lots of people get this process wrong. It's one that is easy to understand but not as easy to write down. These are the important points written in the correct way.

Heat or evaporate the crude oil to make it into a vapour [1]

Cool or condense the hydrocarbons or the fractions [1]
or small molecules at top and / or large molecules at bottom

At different temperatures or boiling points [1]

The alkane methane has the following displayed formula.

0 3 . 2

Draw the displayed formula of propane in the box above.

[2 marks]

(Total 13 marks)

**End**