	1												
0 1	Chlorine gas can be obtained from the electrolysis of sodium chloride solution.												
	The equation for the reaction at the positive electrode is:												
	$2CI_{(aq)} \longrightarrow CI_{2(g)} + 2e^{-}$												
	0.015 moles of gas was collected in 2 minutes.												
0 1 . 1	Calculate the volume (in cm <sup>3</sup> ) of chlorine gas generated in this time. [2 marks]												
	Volume of chlorine gas = cm <sup>3</sup>												
0 2	Ethene reacts with steam to make ethanol in the following reaction:												
	$C_2H_4_{(g)} + H_2O_{(g)} \longrightarrow C_2H_5OH_{(l)}$												
0 2 . 1	What volume of ethene gas would be required to produce 138 grams of ethanol? [3 marks]												
	Volume of ethene gas =(Total 5 marks)												
	End of Question												
	See next page for Data Sheet												

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•	Pe 4	helium 2	20 No	neon 10	40	Ar	argon 18	84	Ϋ́	krypton 36	131	Xe	xenon 54	[222]	Rn	radon 86			
1			19 F	fluorine 9	35.5	ບ	chlorine 17	80	Ъ	bromine 35	127	_	iodine 53	[210]	At	astatine 85			
م			٩C	oxygen 8	32	ა	sulfur 16	62	Se	selenium 34	128	Te	tellurium 52	[209]	Ъо	polonium 84			
ç			7 <b>Z</b>	nitrogen 7	31	٩	hosphorous 15	75	As	arsenic 33	122	sb	antimony 51	209	B	bismuth 83			
4			2 <b>د</b>	carbon 6	28	Si	silicon p 14	73	Ge	jermanium 32	119	Sn	tin 50	207	Ъb	lead 82			
τ Γ			= <b>c</b>	boron 5	27	R	luminium 13	70	Ga	gallium g 31	115	<u>ء</u>	indium 49	204	F	thallium 81			
		I			I		0	65	Zn	zinc 30	112	л С	cadmium 48	201	Нg	mercury 80			
								63.5	5 5	copper 29	108	Ag	silver 47	197	Au	plog 79	[272]	Rg	entgenium 111
								59	ī	nickel 28	106	Pd	alladium 46	195	Ŧ	platinum 78	[271]	Ds	I 10
								59	රී	cobalt 27	103	Ъ	rhodium F	192	<u>_</u>	iridium 77	[268]	Ĕ	eitnerium da 109
	- <b>エ</b>	iydrogen 1						56	Ъе	iron 26	101	Bu	uthenium 44	190	SO	osmium 76	[277]	Hs	hassium m 108
			s			7		55	Чn	anganese 25	[86]	ч Ч	chnetium r 43	186	Be	rhenium 75	[264]	Вh	bohrium 107
			tomic mas	mbol	umber			52	స	hromium m 24	96	в	olybdenum te 42	184	>	74	[266]	Sg	aborgium 1 106
		KEY	relative a	atomic s) name	atomic n			51	>	anadium c 23	63	qN	niobium m 41	181	Ta	antalum t 73	[262]	qO	dubium se
								48	Ē	titanium v 22	16	ц Г	irconium 40	178	Hf	hafnium t 72	[261]	Rf	thfordium 104
		45	š	candium 21	89	~	yttrium z 39	139	La	nthanum 57	[227]	Ac	actinium ru 89						
2			° a	4 4	24	Mg	agnesium 12	40	Ca	calcium s 20	88	ې	trontium 38	137	Ba	barium lau 56	[226]	Ba	radium &
_			- :	3 lithium	23	Na	sodium mi	39	×	otassium (	85	Вb	ubidium st 37	133	S	aesium 55	[223]	ـــــــــــــــــــــــــــــــــــــ	rancium 87
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GAS VOLUMES