Suggest one possible reason why the percentage yield was not $100 \%$.
$\qquad$
$\qquad$

Which reaction has an atom economy of $100 \%$ ?

$$
\mathrm{C}+\mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{CO}_{2}+2 \mathrm{H}_{2}
$$

$$
\mathrm{Cu}+2 \mathrm{HCl} \rightarrow \mathrm{CuCl}_{2}+\mathrm{H}_{2}
$$

$$
\square \quad 2 \mathrm{Cu}+\mathrm{O}_{2} \rightarrow 2 \mathrm{CuO}
$$



Calculate the atom economy for making hydrogen:
[2 marks]

$$
\mathrm{CH}_{4}+\mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{CO}+3 \mathrm{H}_{2}
$$

$\qquad$
$\qquad$

Atom economy = $\qquad$ \%

End of Question


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