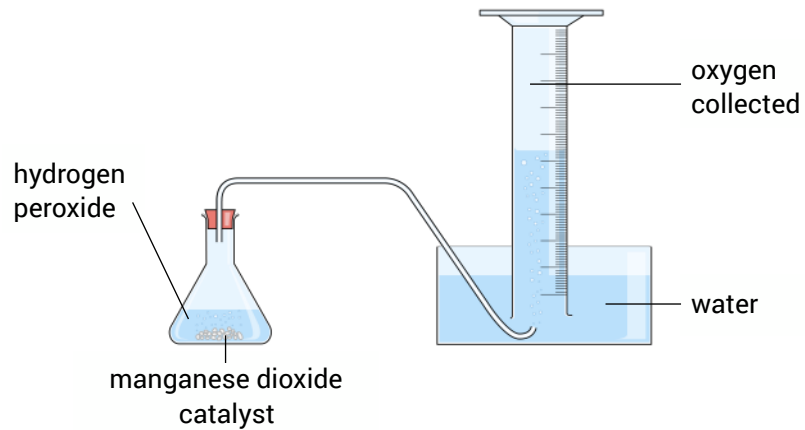
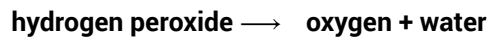


0	1
---	---

The decomposition of hydrogen peroxide solution is often used to make oxygen gas using the apparatus below:



This is the word equation for the reaction:

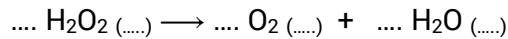


0	1
---	---

1

Complete the balanced symbol equation (including state symbols) for this reaction.

[2 marks]



0	1
---	---

2

If you weighed the conical flask both before and after the reaction, how would you expect the mass at the end of the reaction to compare to the mass at the beginning?

[1 mark]

.....

0	2
---	---

Both metals and non-metals can react with oxygen gas.

Explain, with a reason, how you would expect the mass of the solid reactant to change if the following substances were reacted with oxygen.

0	2
---	---

1

Carbon

[2 marks]

.....

0	2
---	---

2

Calcium

[2 marks]

.....

(Total 7 marks)

End of Question
See next page for Data Sheet

1	2	3	4	5	6	7	0
7 Li lithium 3	9 Be beryllium 4	11 Na sodium 11	12 C carbon 6	14 N nitrogen 7	16 O oxygen 8	19 F fluorine 9	20 Ne neon 10
23 Na sodium 11	24 Mg magnesium 12	27 Al aluminium 13	28 Si silicon 14	31 P phosphorous 15	32 S sulfur 16	35.5 Cl chlorine 17	40 Ar argon 18
39 K potassium 19	40 Ca calcium 20	70 Ga gallium 31	73 Ge germanium 32	75 As arsenic 33	79 Se selenium 34	80 Br bromine 35	84 Kr krypton 36
85 Rb rubidium 37	88 Sr strontium 38	115 In indium 49	119 Sn tin 50	122 Sb antimony 51	128 Te tellurium 52	127 I iodine 53	131 Xe xenon 54
133 Cs caesium 55	137 Ba barium 56	204 Tl thallium 81	207 Pb lead 82	209 Bi bismuth 83	[209] Po polonium 84	[210] At astatine 85	[222] Rn radon 86
[223] Fr francium 87	[226] Ra radium 88	201 Hg mercury 80	201 Hg mercury 80	197 Au gold 79	197 Au gold 79	[272] Rg roentgenium 111	
		59 Ni nickel 28	59 Ni nickel 28	59 Co cobalt 27	59 Co cobalt 27	[271] Ds darmstadtium 110	
		106 Pd palladium 46	106 Pd palladium 46	103 Rh rhodium 45	103 Rh rhodium 45	[268] Mt meitnerium 109	
		190 Os osmium 76	190 Os osmium 76	192 Ir iridium 77	192 Ir iridium 77	[277] Hs hassium 108	
		186 Re rhenium 75	186 Re rhenium 75	186 Re rhenium 75	186 Re rhenium 75	[264] Bh bohrium 107	
		184 W tungsten 74	184 W tungsten 74	184 W tungsten 74	184 W tungsten 74	[266] Sg seaborgium 106	
		181 Ta tantalum 73	181 Ta tantalum 73	181 Ta tantalum 73	181 Ta tantalum 73	[262] Db dubium 105	
		178 Hf hafnium 72	178 Hf hafnium 72	178 Hf hafnium 72	178 Hf hafnium 72	[261] Rf rutherfordium 104	
		139 La lanthanum 57	139 La lanthanum 57	139 La lanthanum 57	139 La lanthanum 57	[227] Ac actinium 89	
		91 Zr zirconium 40	91 Zr zirconium 40	91 Zr zirconium 40	91 Zr zirconium 40		
		93 Nb niobium 41	93 Nb niobium 41	93 Nb niobium 41	93 Nb niobium 41		
		96 Mo molybdenum 42	96 Mo molybdenum 42	96 Mo molybdenum 42	96 Mo molybdenum 42		
		[98] Tc technetium	[98] Tc technetium	[98] Tc technetium	[98] Tc technetium		
		25 Mn manganese	25 Mn manganese	25 Mn manganese	25 Mn manganese		
		26 Fe iron	26 Fe iron	26 Fe iron	26 Fe iron		
		56 Fe iron	56 Fe iron	56 Fe iron	56 Fe iron		
		63.5 Cu copper	63.5 Cu copper	63.5 Cu copper	63.5 Cu copper		
		112 Cd cadmium	112 Cd cadmium	112 Cd cadmium	112 Cd cadmium		
		65 Zn zinc	65 Zn zinc	65 Zn zinc	65 Zn zinc		

1 H hydrogen 1	4 He helium 2
--------------------------------	-------------------------------

KEY

1	→	relative atomic mass
H	→	atomic symbol
hydrogen	→	name
1	→	atomic number