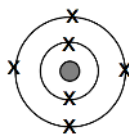


0	1
---	---

The diagrams represent the atoms that make up a molecule of methane. The formula for methane is  $\text{CH}_4$ .



Carbon



Hydrogen

0	1
---	---

1
---

In the box above, draw a molecule of methane. You need only represent the outermost shells in your diagram.

**[2 marks]**

0	1
---	---

2
---

Name the type of bond that you have represented in your diagram.

**[1 mark]**

.....

0	2
---	---

Hydrogen chloride ( $\text{HCl}$ ) can be made by the reaction of hydrogen ( $\text{H}_2$ ) with chlorine ( $\text{Cl}_2$ ).

0	2
---	---

1
---

In the box below, draw a diagram to represent a molecule of hydrogen chloride ( $\text{HCl}$ ).

You need show only the outer energy level (shell) electrons.

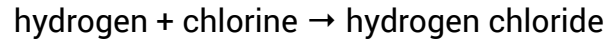
**[3 marks]**

0	2
---	---

 . 

2
---

The word equation for the reaction of hydrogen with chlorine is shown below.



Write a balanced symbol equation for this reaction.

**[2 marks]**

.....

0	3
---	---

One of the gases present in the atmosphere is water vapour.

0	3
---	---

 . 

1
---

Describe the bonding in this compound. (You must include electronic structures in your explanation.)

**[4 marks]**

.....  
.....  
.....  
.....

**(Total 12 marks)**

**End**