| 0 1 | Lithium and other Group 1 metals have different properties from the transition metals. | | |
|---|---|----------------|-------------|
| 0 1 . 1 | Tick (✔) two properties that are properties of transition metals. [2 marks] | | |
| | ☐ They react with o | oxygen. | įz marksj |
| | ☑ They form colour | red compounds. | |
| | ☑ They are strong a | and hard. | |
| | ☐ They have low melting points. | | |
| 0 2 | Copper is a transition metal that is often used to make electrical wiring and pipes for plumbing. | | |
| 0 2 . 1 | Explain why copper is a suitable metal for these uses. [3 r | | [3 marks] |
| | Copper (it) is a good conductor of electricity (to be used in making wires). [1] It can be bent into shape but/and is hard (to be used in making pipes). [1] It does not react with water. [1] | | [1] |
| WARNING: | | | |
| Saying it can be bent into shape alone is not enough as all metals have this property (apart from mercury of course). | | (Tota | ll 5 marks) |