A factory produces steel components for use by a number of different companies. The density of the steel used by the factory is $8000 \mathrm{~kg} / \mathrm{m}^{3}$.


A particular steel cube produced by the factory has a side length of 5 cm .


Calculate the mass of one of these cubes. Write your answer in kilograms.

Mass = $\qquad$ kg


The factory also produces a 10 g steel ball bearing.


Calculate the radius of this ball bearing. Include the unit with your answer.

Remember that the equation for the volume of a sphere is $V=\frac{4}{3} \pi r^{3}$.
$\qquad$ Unit = $\qquad$

A student uses below setup to measure the density of water.


The steps which they take in their investigation are as follows:

- Use measuring cylinder to measure out $200 \mathrm{~cm}^{3}$ of water
- Pour water into beaker
- Turn on and zero electronic balance
- Place beaker containing water onto balance
- Measure mass


Identify the mistake in the method used by the student.

What is the name given to this type of error?


The actual density of water is $1 \mathrm{~g} / \mathrm{cm}^{3}$. Using the incorrect method above, will the student calculate a value for the density of water which is less than or greater than this value? Explain your answer.

